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# Hingganite-(Y) from a syenite pegmatite at Virikkollen, Sandefjord, Vestfold, Norway

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### Introduction

Gadolinite group minerals have the general formula  $A_2X_1Z_2T_2O_8[O_{2-x}(OH)_{2-2x}]$  where  $0 \le x \ge 1$ 1, and  $A = \hat{Y}$ , REE, Ca;  $X = \text{Fe}^{2+}$  or vacancy; Z= Be, B; and T = Si. Several gadolinite group minerals have been found in the Larvik Plutonic Complex: gadolinite-(Ce), gadolinite-(Y), hingganite-(Ce), hingganite-(Y), homilite, and datolite (Larsen 2010). Investigation of pegmatite material collected at the locality Virikkollen near Sandefjord, recently described by Larsen & Kolitsch (2012), revealed a mineral that occurs as pale beige spherulites. Preliminary analyses showed this mineral to be Ca-rich hingganite-(Y), ideally  $Y_2 \square Be_2Si_2O_8(OH)$ , or YBeSiO<sub>4</sub>(OH), which can be expressed as the Fe-free analogue of gadolinite-(Y). The new find initiated a study of this rare mineral.

#### The occurrence

In April 2010, one of the authors (KEL) observed that expansion of the root system of a small rowan tree combined with repeated frost cracking had exposed a cavity about half a metre across in a pegmatite dike in a road cut of Haneholmveien (Haneholm road) on the eastern side of a small hill called Virikkollen. located 1 km SW of the centre of Sandefjord town. The pegmatite dike appears to have a sheet-like shape and is situated about 4-5 m above the street level. Thorough investigation on the locality showed that the primary pegmatite minerals include microcline, black amphiboles, aegirine, magnetite and minor amounts of fluorapatite, zircon, pyrochlore, biotite and albite. The cavity was lined by large aegirine and microcline crystals and hosted a unique assemblage of late stage, hydrothermal minerals (Larsen & Kolitsch 2012), among them zektzerite and an aspedamite-like mineral (UK-17). Other hydrothermal minerals included second generation aegirine, bastnäsite-(Ce), bertrandite, chalcedony, epididymite, ilmenite, pale mica, milarite, monazite-(Ce), hingganite-(Y), opal, second generation of pyrochlore, quartz, second generation of zircon, plus black manganese oxide crusts and unspecified clay material.

## The mineral description

Hingganite-(Y) from the Virikkollen locality occurs as pale beige, radiating spherulites up to 1 mm across, often as intergrown aggregates. The spherulites are abundantly dispersed in the grundmass, mainly quartz and aegirine. The streak is white. No fluorescence is observed under short- and long-wavelength UV light. A back-scatter image of a split and polished spherulite (Fig. 1) show that the composition is rather complex with intergrown silicates, and probably a fine alteration along the hingganite needles. In addition, the spherulite is rimmed by a late-stage hingganite with slightly different chemical composition.

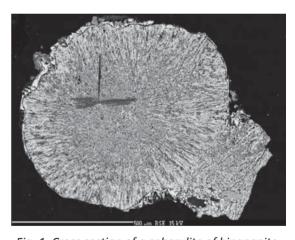


Fig. 1. Cross section of a spherulite of hingganite-(Y) from Virikkollen, Sandefjord. SEM micrograph (BSE mode), R. Škoda.

### The chemical composition

Quantitative analyses were carried out using an electron microprobe. The average of 9 spot analyses on the main part of the mineral is shown in Table 1. Ba, Sr, P, Sc, K, Mg, Na and Al were sought, but not detected. The quadrant diagram of the main species in the gadolinite group (Fig. 2) shows the distribution of the apfu of (Fe+Mn) vs. (Y+REE), clearly indicating that the Virikkollen mineral as a hingganite, and

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Table 1. Chemical composition determined by LA-ICP-MS and EMP (average of 9 analyses), and number of atoms calculated on the basis of Si = 2.

SiO2         27.13 (26.81-27.29)         2.000           ThO2         0.13 (0.09-0.14)         0.002           UO2         0.02 (0.00-0.06)         0.000           Y2O3         31.22         30.60 (29.33-31.58)         1.200           La2O3         0.13         0.14 (0.09-0.24)         0.004           Ce2O3         1.27         1.37 (1.24-1.51)         0.037           Pr2O3         0.29         0.40 (0.33-0.45)         0.011           Nd2O3         2.33         2.60 (1.91-2.83)         0.068           Sm2O3         1.90         1.40 (1.15-1.67)         0.036           Eu2O3         0.24         0.19 (0.15-0.24)         0.005           Gd2O3         3.70         3.49 (3.30-3.67)         0.085           Tb2O3         0.71         0.65 (0.58-0.76)         0.016           Dy2O3         4.55         4.07 (3.80-4.31)         0.097           Ho2O3         0.94         0.85 (0.83-0.96)         0.020           Er2O3         2.40         2.13 (2.03-2.27)         0.049           Tm2O3         0.25         0.21 (0.16-0.26)         0.005           Yb2O3         1.27         0.82 (0.70-0.95)         0.019           Lu2O3         0.10 <th></th> <th>LA-ICP-MS</th> <th>EMP (<i>n</i>=9)</th> <th>No. of</th>		LA-ICP-MS	EMP ( <i>n</i> =9)	No. of
ThO <sub>2</sub> 0.13 (0.09-0.14) 0.002  UO <sub>2</sub> 0.02 (0.00-0.06) 0.000  Y <sub>2</sub> O <sub>3</sub> 31.22 30.60 (29.33-31.58) 1.200  La <sub>2</sub> O <sub>3</sub> 0.13 0.14 (0.09-0.24) 0.004  Ce <sub>2</sub> O <sub>3</sub> 1.27 1.37 (1.24-1.51) 0.037  Pr <sub>2</sub> O <sub>3</sub> 0.29 0.40 (0.33-0.45) 0.011  Nd <sub>2</sub> O <sub>3</sub> 2.33 2.60 (1.91-2.83) 0.068  Sm <sub>2</sub> O <sub>3</sub> 1.90 1.40 (1.15-1.67) 0.036  Eu <sub>2</sub> O <sub>3</sub> 0.24 0.19 (0.15-0.24) 0.005  Gd <sub>2</sub> O <sub>3</sub> 3.70 3.49 (3.30-3.67) 0.085  Tb <sub>2</sub> O <sub>3</sub> 0.71 0.65 (0.58-0.76) 0.016  Dy <sub>2</sub> O <sub>3</sub> 4.55 4.07 (3.80-4.31) 0.097  Ho <sub>2</sub> O <sub>3</sub> 0.94 0.85 (0.83-0.96) 0.020  Er <sub>2</sub> O <sub>3</sub> 2.40 2.13 (2.03-2.27) 0.049  Tm <sub>2</sub> O <sub>3</sub> 0.25 0.21 (0.16-0.26) 0.005  Yb <sub>2</sub> O <sub>3</sub> 1.27 0.82 (0.70-0.95) 0.019  Lu <sub>2</sub> O <sub>3</sub> 0.10 0.06 (0.03-0.08) 0.001  CaO 4.27 (3.77-4.69) 0.338  MnO 0.10 (0.04-0.14) 0.006  FeO 2.60 (2.46-2.82) 0.161  BeO 8.79 9.33 (9.12-9.51)* 1.652*  B <sub>2</sub> O <sub>3</sub> 3.78 2.65 (2.34-2.91)* 0.338*  H <sub>2</sub> O 3.47 (2.94-3.75)** 1.706**				atoms
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	SiO <sub>2</sub>		27.13 (26.81-27.29)	2.000
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$ThO_{\scriptscriptstyle{2}}$		0.13 (0.09-0.14)	0.002
La <sub>2</sub> O <sub>3</sub> 0.13       0.14 (0.09-0.24)       0.004         Ce <sub>2</sub> O <sub>3</sub> 1.27       1.37 (1.24-1.51)       0.037         Pr <sub>2</sub> O <sub>3</sub> 0.29       0.40 (0.33-0.45)       0.011         Nd <sub>2</sub> O <sub>3</sub> 2.33       2.60 (1.91-2.83)       0.068         Sm <sub>2</sub> O <sub>3</sub> 1.90       1.40 (1.15-1.67)       0.036         Eu <sub>2</sub> O <sub>3</sub> 0.24       0.19 (0.15-0.24)       0.005         Gd <sub>2</sub> O <sub>3</sub> 3.70       3.49 (3.30-3.67)       0.085         Tb <sub>2</sub> O <sub>3</sub> 0.71       0.65 (0.58-0.76)       0.016         Dy <sub>2</sub> O <sub>3</sub> 4.55       4.07 (3.80-4.31)       0.097         Ho <sub>2</sub> O <sub>3</sub> 0.94       0.85 (0.83-0.96)       0.020         Er <sub>2</sub> O <sub>3</sub> 2.40       2.13 (2.03-2.27)       0.049         Tm <sub>2</sub> O <sub>3</sub> 0.25       0.21 (0.16-0.26)       0.005         Yb <sub>2</sub> O <sub>3</sub> 1.27       0.82 (0.70-0.95)       0.019         Lu <sub>2</sub> O <sub>3</sub> 0.10       0.06 (0.03-0.08)       0.001         CaO       4.27 (3.77-4.69)       0.338         MnO       0.10 (0.04-0.14)       0.006         FeO       2.60 (2.46-2.82)       0.161         BeO       8.79       9.33 (9.12-9.51)*       1.652*	$UO_2$		0.02 (0.00-0.06)	0.000
Ce2O3       1.27       1.37 (1.24-1.51)       0.037         Pr2O3       0.29       0.40 (0.33-0.45)       0.011         Nd2O3       2.33       2.60 (1.91-2.83)       0.068         Sm2O3       1.90       1.40 (1.15-1.67)       0.036         Eu2O3       0.24       0.19 (0.15-0.24)       0.005         Gd2O3       3.70       3.49 (3.30-3.67)       0.085         Tb2O3       0.71       0.65 (0.58-0.76)       0.016         Dy2O3       4.55       4.07 (3.80-4.31)       0.097         Ho2O3       0.94       0.85 (0.83-0.96)       0.020         Er2O3       2.40       2.13 (2.03-2.27)       0.049         Tm2O3       0.25       0.21 (0.16-0.26)       0.005         Yb2O3       1.27       0.82 (0.70-0.95)       0.019         Lu2O3       0.10       0.06 (0.03-0.08)       0.001         CaO       4.27 (3.77-4.69)       0.338         MnO       0.10 (0.04-0.14)       0.006         FeO       2.60 (2.46-2.82)       0.161         BeO       8.79       9.33 (9.12-9.51)*       1.652*         B <sub>2</sub> O3       3.78       2.65 (2.34-2.91)*       0.338*         H <sub>2</sub> O       3.47 (2.94-3.75)**	$Y_2O_3$	31.22	30.60 (29.33-31.58)	1.200
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$La_2O_3$	0.13	0.14 (0.09-0.24)	0.004
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Ce <sub>2</sub> O <sub>3</sub>	1.27	1.37 (1.24-1.51)	0.037
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Pr_2O_3$	0.29	0.40 (0.33-0.45)	0.011
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Sm_2O_3$	1.90	1.40 (1.15-1.67)	0.036
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Eu_2O_3$	0.24	0.19 (0.15-0.24)	0.005
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Gd_2O_3$	3.70	3.49 (3.30-3.67)	0.085
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Tb_2O_3$	0.71	0.65 (0.58-0.76)	0.016
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Dy <sub>2</sub> O <sub>3</sub>	4.55	4.07 (3.80-4.31)	0.097
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Ho <sub>2</sub> O <sub>3</sub>	0.94	0.85 (0.83-0.96)	0.020
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\mathrm{Er_2O_3}$	2.40	2.13 (2.03-2.27)	0.049
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\text{Tm}_{2}\text{O}_{3}$	0.25	0.21 (0.16-0.26)	0.005
CaO	$Yb_2O_3$	1.27	0.82 (0.70-0.95)	0.019
MnO       0.10 (0.04-0.14)       0.006         FeO       2.60 (2.46-2.82)       0.161         BeO       8.79       9.33 (9.12-9.51)*       1.652*         B <sub>2</sub> O <sub>3</sub> 3.78       2.65 (2.34-2.91)*       0.338*         H <sub>2</sub> O       3.47 (2.94-3.75)**       1.706**	Lu <sub>2</sub> O <sub>3</sub>	0.10	0.06 (0.03-0.08)	0.001
FeO       2.60 (2.46-2.82)       0.161         BeO       8.79       9.33 (9.12-9.51)*       1.652*         B2O3       3.78       2.65 (2.34-2.91)*       0.338*         H2O       3.47 (2.94-3.75)**       1.706**	CaO		4.27 (3.77-4.69)	0.338
BeO       8.79       9.33 (9.12-9.51)*       1.652*         B <sub>2</sub> O <sub>3</sub> 3.78       2.65 (2.34-2.91)*       0.338*         H <sub>2</sub> O       3.47 (2.94-3.75)**       1.706**	MnO		0.10 (0.04-0.14)	0.006
B <sub>2</sub> O <sub>3</sub> 3.78 2.65 (2.34-2.91)* 0.338* H <sub>2</sub> O 3.47 (2.94-3.75)** 1.706**	FeO		2.60 (2.46-2.82)	0.161
H <sub>2</sub> O 3.47 (2.94-3.75)** 1.706**	BeO	8.79	9.33 (9.12-9.51)*	1.652*
2	$B_2O_3$	3.78	2.65 (2.34-2.91)*	0.338*
Total 98.69	$H_2O$		3.47 (2.94-3.75)**	1.706**
	Total		98.69	

<sup>\*</sup> Be and B were calculated on the basis of gadolinite-hingganite and datolite end member composition, respectively.

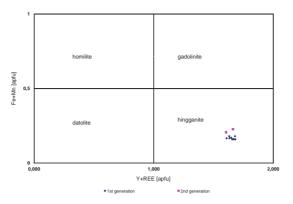


Fig. 2. Quadrant diagram on the apfu of (Fe+Mn) vs. (Y+REE) of the main species in the gadolinite group showing the position of the spot EMP analyses, all within the hingganite region.

with Y as the dominating rare earth element the mineral is hingganite-(Y). The empirical formula for hingganite-(Y) from Virikkollen on the basis of Si = 2 is

 $\begin{array}{l} (Y_{1.20} Ca_{0.34} Dy_{0.10} Gd_{0.09} Nd_{0.07} Er_{0.05} Sm_{0.04} Ce_{0.04} \\ Ho_{0.02} Yb_{0.02} Tb_{0.02} Pr_{0.01})_{\Sigma 1.99} (\square_{0.84} Fe_{0.16})_{\Sigma 2.00} \\ (Be_{1.65} B_{0.34})_{\Sigma 1.99} Si_{2.00} O_{10.00} (OH)_{1.68}. \end{array}$ 

Two spot analyses on the outer zone of late stage hingganite-(Y) show that this phase is slightly enriched in Fe (average 0.22 apfu) (Fig 2).

The contents of Be and B were calculated on the basis of gadolinite-hingganite and datolite end member composition, respectively. In addition, REE, Be and B were determined using a laser ablation inductively coupled plasma mass spectrometry instrument (LA-ICP-MS). The results are shown in Table 1. Most of the analytical results for the REE by this technique are within the variations of the EMP analyses. Interestingly, the calculated Be and B contents differ only slightly from that measured by LA-ICP-MS.

# X-ray crystallography

X-ray powder diffraction data on the hingganite-(Y) were obtained using a Panalytical X'pert Pro diffractometer equipped with automatic divergence slits ( $CuK\alpha_1$  radiation,  $\lambda = 1.54056$  Å). Finely ground mineral was dispersed on a zero-background silicon plate. Data were collected from 5° to 70° 20. Lanthanum hexaboride ( $LaB_6$ , NIST

SRM 660a) was used as internal standard. The X-ray powder diffraction pattern was indexed using the monoclinic space group  $P2_1/a$ , and least-squares refinement was done by the program CELREF (Laugier & Bochu 1999). The unit-cell dimensions found are a = 9.8654(17), b = 7.6033(12), c = 4.7561(7) Å,  $\theta = 90.11(3)^\circ$ , V = 356.75(10) Å<sup>3</sup>.

## Conclusion

The diversity of minerals in the pegmatites in the Larvik Plutonic Complex (LPC) became famous through the monograph by Brøgger (1890). Among all the alkaline complexes worldwide, the pegmatites of the LPC show the largest diversity in beryllium minerals with 26 species, among them 11 type species (Raade 2008; Larsen 2010). Hingganite-(Y) has previously been identified from the LPC: as tiny, beige crystals in the Tuften quarry, Tvedalen, and as groups of radiating, pale rose coloured crystals at Bratthagen, Lågendalen. The present paper is yet another description of the rare beryllium mineral hingganite-(Y) from the LPC.

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Detail of the Virikkollen sample showing white spherulites of hingganite-(Y) associated with quartz, aegirine and Fe-hydroxides. Field of view 20 mm.

Photo by A. O. Larsen.

<sup>\*\*</sup> OH was calculated to keep the formula electro-neutral.